



cket No.: ACO2849US

DECLARATION AND POWER OF ATTORNEY

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled: CONDUCTIVE COATING COMPOSITION

the specification of which:

☒ is attached hereto, and/or

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above. All factual statements made in the specification of my own knowledge are true and all factual statements made on information and belief are believed to be true.

I acknowledge the duty to disclose information which is material to the patentability of this application in accordance with Title 37, Code of Federal Regulations, Sec. 1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code, Sec. 119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Priority Claimed

<u>01200314.1</u>	<u>EUROPE</u>	<u>29 JANUARY 2001</u>	<u>X</u> Yes	<u>    </u> No
(Number)	(Country)	(Day/Month/Year)		

I hereby claim the benefit under Title 35, United States Code § 119 of any provisional application(s) listed below:

Appln. Ser. No.	Country	Day/Month/Year
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OFFICE OF PETITIONS

I hereby claim the benefit under Title 35, United States Code, Sec. 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, Sec. 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, Sec. 1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

(Appln. Ser. No.)	(Filing Date)	(Status: patented, pending, abandoned)
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**POWER OF ATTORNEY:** As a named inventor, I hereby appoint the following as my attorneys of record, with full power of substitution and revocation, to prosecute this application and to transact all business in the Patent Office:

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further, that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Sec. 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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## Conductive coating composition

This patent application claims priority of European patent application No. 01200314.1, filed January 29, 2001.

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### Background of the Invention

The present invention relates to a conductive and/or anti-static coating composition. Conductive coatings can be used for making conductive films, e.g., for making conductor circuits on printed circuit boards or integrated circuits, or for the shielding  
10 against electromagnetic waves, e.g. the shielding of computer apparatus against external electromagnetic interference.

Conductive and/or anti-static coating compositions are known in the art.

15

US 5,447,791 relates to the preparation of conductive small particles possessing high conductivities. Such micron and sub-micron particles can be used as color toner particles and in conductive powder coatings. In this process a conductive filler is normally added at the start of the polymerisation process. Carbon black and  
20 metal oxides such as iron oxides, titanium oxide, tin dioxide and metal powders are used as conductive filler.

WO 99/39839 discloses an electrically conductive powder coating containing a transparent or white conductive powder such as zinc oxide powder or tin dioxide  
25 powder doped with antimony or an inorganic powder coated with tin dioxide powder doped with antimony. The coating film has a volume specific resistance in the range of  $10^7$  to  $10^{13} \Omega \cdot \text{cm}$ .

In EP 425 677 an electrically conductive coating is disclosed comprising an  
30 electron-beam curable resin and an electrically conductive fine powder having an average particle diameter of 1 to 50  $\mu\text{m}$ . The electrically conductive powder can be a metallic powder, e.g. gold, silver, copper, nickel, chromium, palladium, aluminium,

tungsten, molybdenum or platinum, inorganic or organic powders coated with these metallic powders, powders of metal oxide, carbon black or graphite. Conductive powder coatings are not disclosed in this publication.

5 WO 96/06437 relates to electroconductive powders based on tin oxide doped with tantalum, niobium or phosphorus that can be used in coatings. The powders can be used to prepare transparent, conductive films and impart electrical conductivity and anti-static properties.

10 In WO 97/32936 a powder coating is disclosed comprising zinc powder, zinc flakes and/or graphite having a minimum conductivity of  $3 \Omega/\text{cm}^2$ .

15 In US 5,556,576 conductive polymeric coatings are disclosed comprising a fine conductive powder, for example, metal powders, metal coated mircobeads of glass, ceramics and carbon black or graphite of appropriate particle size. Conductive powder coatings are not disclosed in this publication.

20 In JP 62-197473 a solventborne coating composition comprising a synthetic resin and metal fibers is disclosed. In view of present day legislation, solventborne coatings are less desired in view of the non VOC-compliant solvents that are present in these coating compositions. Conductive powder coatings are not disclosed in this publication.

25 In JP 62-018481 a solventborne coating composition comprising a synthetic resin and metal fiber fillers is disclosed. In view of present day legislation, solventborne coatings are less desired in view of the non VOC-compliant solvents that are present in these coating compositions. Conductive powder coatings are not disclosed in this publication.

30 In JP 61-157541 a conductive filler is disclosed comprising glass fibres that are chemically coated with nickel. The filler is used in combination with ABS resin to

prepare conductive housing for electric appliances. Conductive powder coatings are not disclosed in this publication.

5 In EP 0 367 224 an electroconductive coating material is disclosed containing a conductive organic fiber. Metallic fibres and conductive powder coatings are not disclosed in this publication.

A major disadvantage of coating compositions known in the art is that either they can not be used as white coatings or that they cannot be used as powder coatings.

10 Normally, the conductive particles that are present in the composition give the composition a grey or greyish appearance, in particular when carbon black or graphite is used as conductive material in the coating, but also when metals or metal oxides as, for example, zinc or zinc oxide are used, after some time a coating with a greyish appearance is obtained. For a number of applications the use of  
15 conductive powder coatings with a white appearance is highly desirable.

Surprisingly, it was found that white powder coatings with conductive and/or anti-static properties can be obtained from powder coating compositions comprising metallic fibres. In particular good results are obtained in view of long term stability of  
20 the colour of the coating when a powder coating composition comprising stainless steel fibres is used.

#### Summary of the Invention

25 In one embodiment, the invention relates to electrically conductive powder coating compositions having anti-static properties comprising metallic fibres.

In another embodiment, the invention relates to powder coating compositions having stainless steel fibres.

### Detailed Description of the Invention

For the purpose of the present invention, a fibre is an elongated object with dimensions in two perpendicular directions that are more or less the same and a dimension in the third direction, perpendicular to the other two perpendicular directions, that is larger than in the other two directions, i.e. an object having a certain length and a diameter. If the diameter of the fibre is not constant over its length, the diameter D of the fibre equals the average diameter over the whole length (L) of the fibre. When the fibre does not have a circular cross section, the diameter of the fibre equals  $2 \times \text{SQRT}(\text{cross section}/\pi)$ .

Within the framework of the present invention, metallic fibres include conductive fibres containing metal(s) and/or metal alloy(s), for example, metal fibres containing only one metal or metal-coated non-conductive fibres.

To be able to have a coating with a smooth surface, these fibres should not be too long, but on the other hand, to have good conductive and/or anti-static properties, the fibres should not be too short. It was found that an optimum in these properties is obtained by using metallic fibres with an L/D ratio in the range between 5 and 75, wherein L is the length of the fibre and D is the diameter of the fibre. The diameter of the metallic fibres should normally be in the range from 5 to 25  $\mu\text{m}$ .

To have a coating with good mechanical properties and appearance, the concentration of the metallic fibres in the coating composition should not be too high. To have an optimum between mechanical properties, appearance and conductive and/or anti-static properties, it was found that the content of metal fibres in the coating composition should be in the range of 0.5 to 20 wt.% based upon the total solids weight in the coating composition. Preferably the content of metallic fibres is in the range of 5 to 15 wt.% based upon the total solids weight in the coating composition.

In EP 286 168 granular plastic composites are disclosed containing metal fibers. These composites can be used for the shaping of thermoplastic articles with

antistatic properties or shielding properties against electromagnetic radiation. In order to obtain a reasonable shielding efficiency with a low metallic fiber content (< 1% by weight) it is important to maintain in the composite a relatively high L/D ratio for the fibers of  $\geq 100$ , preferably in the range between 7100 and 2000. In this publication no reference is made to the use of metallic fibres in coatings and the L/D ratio is much larger than the optimum L/D ratio for metallic fibres used in the coating composition according to the present invention.

For larger scale production in an industrial environment, preference is given to the use of coating compositions that are substantially solvent free, e.g., powder coating compositions.

It was found that metallic fibres can be very easily mixed with a normal white powder coating to get a conductive and/or anti-static powder coating composition. However, it is also possible to add the metallic fibres during the manufacturing process of the powder coating composition.

Surprisingly, the presence of the metallic fibres does not have any negative effect on the application of the powder coating via conventional techniques whereby the powder coating particles are electrostatically charged.

In principle any combination known to the man skilled in the art of binders, crosslinkers, additives, fillers, pigments, etc. that are used in known non-conductive powder coating compositions can be used in the conductive powder coating composition comprising metallic fibres according to the present invention. Both thermal curable and radiation curable powder coating compositions can be used.

The conductive powder coating composition according to the present invention can be prepared by mixing and milling all components until a composition with the desired particle size is obtained. Normally, the composition is mixed to the extent that all particles in the composition have a size below 150  $\mu\text{m}$ . However, after the



milling of the composition it is normally not sieved, since this could result in the removal of the conductive fibres.

The invention will be elucidated with reference to the following examples. These are intended to illustrate the invention but are not to be considered as limiting in any manner the scope thereof.

#### Examples

##### Conductivity Measurement Procedure

10 The conductivity of the coating layers was measured using a UNILAP ISO X NORMA Insulation Tester, which is a calibrated 1000Vdc closed circuit meter.

For the measurements two electrodes are used. Each electrode consists of a brass cube of 50mm sides and a total weight of 1 kg. A square rubber pad of 5mm thickness, which has been covered with a thin metal foil, is fixed on the underside  
15 to give maximum contact between electrode and substrate.

To measure the conductivity the electrodes are placed onto the same face of the panel spaced apart and a voltage is applied – typically 100Vdc. The resistance between the two electrodes is then measured. The more conductive the coating, the lower the resistance. In each case, the applied voltage is quoted alongside the  
20 measured resistance.

In the examples the following components were used in the preparation of the coating compositions:

<u>COMPONENT</u>	<u>TYPE</u>	<u>SUPPLIER</u>
ACEMATT TS 100	dry flow additive	Degussa
Additol VXL9890	catalyst masterbatch	Solutia
ALFTALAT 03580	polyester resin	Vianova
ALFTALAT 03872	polyester resin	Vianova
ALFTALAT AN720	polyester resin	Vianova
ARALDITE GT 7004	epoxy resin	Ciba
Barytes	filler	Viaton Industries
Beki-shield GR90	stainless steel fibres	Bekaert Fibre Technologies
Benzoin	anti-gassing agent	Caffaro
BYK-360 P	flow aid	BYK Chemie
CAB.551.02	texturing agent	Eastman Chemical
CASAMID 1457	curing agent / hardener	Thomas Swan
D.E.R. 663U-E epoxy	epoxy resin	Dow
D.E.R. 692 epoxy	epoxy resin	Dow
G.LO 420 iron oxide	pigment	Bayer
IRGACURE 184	initiator	Ciba
Irganox 1010	anti-oxidant	Ciba
Licowax PE 520 P	polyethylene wax	Clariant
MICRODOL H600	extender / filler	Norwegian Talc Minerals
Titanium Dioxide	pigment	Millenium
URACROSS P3125	polyester resin	DSM
URACROSS P3307	curing agent / hardener	DSM
URALAC P5170	polyester resin	DSM
VESTAGON B 68	hardener / matting agent	Creanova
VULCAN XC72	conductive black pigment	Cabot

Example 1 (White epoxy-polyester coating over steel)

- Two compositions (1 and 2) were prepared by melt compounding of the following
- 5 ingredients using an extruder:

Ingredient	Composition 1 pbw	Composition 2 pbw
ALFTALAT AN720		360
ALFTALAT 03580	360	
D.E.R. 692	240	240
Additol VXL9890		40
Licowax PE 520 P	2	2
Irganox 1010	2	2
Benzoin	2	2
Barytes	254.20	115
Stainless Steel Fibres		100
Titanium Dioxide	130	130
Black Pigment	3.5	3.5
G.LO 420 iron oxide	5	5
CAB.551.02	0.6	0.5

An electrically conductive powder coating composition was prepared by melt compounding 90 pbw (parts by weight) of composition 1 and 10 pbw of composition 2 followed by milling of the extrudate to have a particle size of 100% below 150  $\mu\text{m}$ . The powder coating was not sieved, as would usually be the case, since this may have resulted in the removal of the fibres. The powder coating according to the invention was electrostatically sprayed onto mild steel panels and stoved (15 minutes at 180° C) to give a white coating of 80  $\mu\text{m}$  applied film thickness having a 60° gloss of 33%.

For comparison, a conductive powder coating according to the state of the art was prepared using the following ingredients:

Ingredient	Composition 4 (pbw)
Polyester URALAC P5170	21.7
CASAMID 1457	2.4
Benzoin	0.3
Additol VXL9890	2.2
BYK-360 P	0.9
VULCAN XC72	7.2
Titanium Dioxide	7.2
ARALDITE GT 7004	36.2
MICRODOL H600	21.7
ACEMATT TS 100	0.2

Composition 5 is a commercial product (RESICOAT 507901), available from Akzo Nobel Powder Coatings (Germany).

5 Compositions 4 and 5 were also applied to steel panels and cured at elevated temperature.

The conductivity of the coated substrates was measured according to the procedure given above.

10 In addition to the coated substrates, the conductivity of some non-coated metal substrates was also measured. The results of these test is shown in Table 1

Table 1; Conductivity Testing at 100 V

Sample	Conductivity ( $\Omega$ )
Composition 3	22
Composition 4*	4400
Composition 5*	7300
Tinplate steel*	2
Untreated steel*	2
ZnPO <sub>4</sub> steel*	170
FePO <sub>4</sub> steel*	10
Untreated aluminium*	2
Chromated aluminium*	3

\*) Comparison.

15

It can be seen from the conductivity testing that the powder coating according to the invention has a surface resistance that is 200 times lower than the best available according to the prior art (an average of 22  $\Omega$  for Composition 3, compared to an average of 4400  $\Omega$  for Composition 4).

20

#### Example 2 (White epoxy-polyester coating over steel)

A conducting powder coating composition was prepared by using the following components:

Ingredient	Composition 6 pbw
D.E.R. 663U-E epoxy	423
BYK-360 P	11
Benzoin	6
Black Pigment	0.37
Blue Pigment	0.86
Yellow Pigment	0.46
VESTAGON B 68	38
MICRODOL H600	58
Titanium Dioxide	346
ALFTALAT 03872 / Steel Fibre Masterbatch	100
ALFTALAT 03872	15
ACEMATT TS 100	1

The Steel Fibre Masterbatch was prepared by hot melt compounding of the ALFTALAT polyester resin and steel fibres (Beki-shield GR90) (10:1 w/w) using an extruder. The resulting extrudate was crushed and used as an ingredient for composition 6. Composition 6 was prepared as a powder coating in the same manner as composition 3. Analysis of the powder coating composition revealed that the composition comprises steel fibres with an L/D ratio in the range of 10 – 50 at a fibre diameter of approximately 10 µm.

#### Example 3 (Clear UV curable coating over MDF)

Some clear, UV-curable powder coatings were prepared by using the following ingredients:

Ingredient	Composition 7 pbw	Composition 8 pbw
URACROSS P3125	830	730
URACROSS P3125 / Steel Fibre Masterbatch		100
URACROSS P3307	170	170
IRGACURE 184	10	10
BYK-360 P	10	10

The Steel Fibre Masterbatch for composition 8 was manufactured in the same manner as that used for composition 6 using the URACROSS polyester resin. Compositions 7 and 8 were prepared as powder coatings in the same manner as composition 3. Compositions 7 and 8 were applied to MDF and cured using a Beltron IR / UV oven at 0.3m/min track speed, 66 / 66 / 0 IR zone dial settings.

Example 4 (Clear UV curable coating over polycarbonate)

Compositions 7 and 8 were applied to a polycarbonate substrate and cured using a Beltron IR / UV oven at 0.6m/min track speed, 80 / 70 / 20 IR zone dial settings.

Table 2: Conductivity testing at 100V

Sample	Conductivity ( $\Omega$ )
Composition 6	16
Interzinc 52HS (panel 1)	54
Interzinc 52HS (panel 2)	26
Polyzinc 660	3 - 8

Notes: Interzinc 52HS was applied to steel panels.  
Polyzinc 660 was applied to untreated aluminium panels.

Interzinc 52HS is a high solids, low VOC, zinc-rich liquid epoxy primer available from Akzo Nobel International Coatings (UK) Protective Coatings.

Polyzinc 660 is available from Akzo Nobel Powder Coatings (France).

Composition 6 has a conductivity comparable with the zinc-rich coatings (Interzinc and Polyzinc), yet is white in colour while the others are limited to grey.

Composition 7 and 8 were applied over insulating substrates. The substrate coated with composition 8 showed improved EMF shielding in comparison to composition

7.

## Claims

1. Electrically conductive powder coating composition having anti-static properties comprising metallic fibres.
- 5 2. Powder coating composition according to claim 1 wherein the metallic fibres have an L/D ratio in the range between 5 and 75, wherein L is the length of the fibre and D is the diameter of the fibre.
- 10 3. Powder coating composition according to claim 1 wherein the metallic fibres have a diameter in the range from 5 to 25  $\mu\text{m}$ .
4. Powder coating composition according to claim 1 wherein the metallic fibres are stainless steel fibres.
- 15 5. Powder coating composition according to claim 1 wherein the content of metal fibres in the composition is in the range of 0.5 to 20% by weight based upon the total solids weight of the coating composition.
- 20 6. Powder coating composition according to claim 5 wherein the content of metal fibres in the composition is in the range of 5 to 15 % by weight.
7. Powder coating composition according to claim 2 wherein the metallic fibres have a diameter in the range from 5 to 25  $\mu\text{m}$ .
- 25 8. Powder coating composition according to claim 2 wherein the metallic fibres are stainless steel fibres.
9. Powder coating composition according to claim 3 wherein the metallic fibres are stainless steel fibres.

10. Powder coating composition according to claim 2 wherein the content of metal fibres in the composition is in the range of 0.5 to 20% by weight based upon the total solids weight of the coating composition.
- 5 11. Powder coating composition according to claim 3 wherein the content of metal fibres in the composition is in the range of 0.5 to 20% by weight based upon the total solids weight of the coating composition.
- 10 12. Powder coating composition according to claim 4 wherein the content of metal fibres in the composition is in the range of 0.5 to 20% by weight based upon the total solids weight of the coating composition.
- 15 13. Powder coating composition according to claim 10 wherein the content of metal fibres in the composition is in the range of 5 to 15 % by weight.
14. Powder coating composition according to claim 11 wherein the content of metal fibres in the composition is in the range of 5 to 15 % by weight.
- 20 15. Powder coating composition according to claim 12 wherein the content of metal fibres in the composition is in the range of 5 to 15 % by weight.



Abstract of the Invention

The invention relates to an electrically conductive powder coating composition having anti-static properties comprising metallic fibres. Further, the coating composition shows effective EMF shielding. Very good results are found when the  
5 metallic fibres have an L/D ratio in the range between 5 and 75, wherein L is the length of the fibre and D is the diameter of the fibre. In a preferred embodiment stainless steel fibres are incorporated in the powder coating composition.